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**Allocation of Papers in Section C
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Section C of *Acta Crystallographica* will be divided from Volume 43 into the following formal categories for the convenience of its readers:

Inorganic compounds

Those not containing any organic (C_xH_y) or substituted organic ($C_xH_{y-1}F$, C_xF_y etc.) moiety. Included are metal

elements, intermetallic compounds, non-metal elements, compounds of non-metals (excluding organic compounds as defined above), salts of inorganic acids, coordination complexes not containing organic moieties. Metal carbides, carbonates, cyanides, isocyanides, cyanates, isocyanates, thiocyanates are regarded as inorganic, as are coordination complexes with carbon present only as carbonyl, cyanide, cyanato etc. ligands.

Examples: Be, Gd_2Pt , C(graphite), SiC, $NaHCO_3$, $K[Co(CN)(CO)_2(NO)]$.

Organometallic compounds

Those containing one or more metal atom and one or more organic (as defined above) moiety (not simply a metal–carbon bond). Included are metal salts of organic acids, coordination complexes in which carbon is present in an organic or substituted-organic ligand, metal(or metal salt)–organic addition compounds. Al, Ge, As, Sb and Po are defined as metals.

Examples: $Ba(CHO_2)_2$, $K_4[Zr(C_2O_4)_4] \cdot 5H_2O$
(four oxalate ligands),
 $K[PtCl_3(C_2H_4)]$, $CaBr_2 \cdot 4CH_4O$.

Organic compounds

Those containing carbon and hydrogen (or hydrogen partially or fully substituted by a non-metal) with or without additional non-metals. B, Si, Se, Te and At are defined as non-metals.

Examples: $C_7H_4N_4$, CCl_4 (carbon tetrachloride or tetra-chloromethane),
 C_6N_4 (ethylenetetracarbonitrile), $C_4H_8Se_4$.

Biological macromolecules

Any substance containing a polypeptide, polynucleotide or polysaccharide chain longer than 25 residues.

Short-format papers will be divided into the same categories as for full papers.